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elements, underline the one that would replace the other element in a compound. Teaching Transparency Master 2 Answer Key Buttup Section Focus Transparency 34 and Master Teach Problem ...

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with Chapter 7, Section 7.2 1. How many valence electrons does a neutral magnesium (Mg) atom have? 2. What is the charge on a magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so? 3. How many valence electrons does a single neutral chlorine (Cl) atom have? 4.

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6. Considering your answer to question 5, how many electrons can  
fit in the entire 5<sup>th</sup> energy level? can (A<sub>Vel</sub> is a ha AC5a) 5<sup>0</sup>đ 7.  
Write the notation for an electron spinning clockwise in a p sublevel  
in the 4<sup>th</sup> energy level.



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1. What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 5.2  
2. What is the relationship between the size of an s orbital and the principal energy level in which it is found?  
3. What is the shape of a p orbital? How many p orbitals are there in a

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level? 4.

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