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variables, such as the
temperature and
volume. Most of the
aggregate
thermodynamic
variables of the
system, such as the
total energy ...

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Overview:

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Thermodynamics is

the study of the

energy associated

with temperature.

Thermodynamics

(from the Greek

words meaning...

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branch of physics that deals with heat, work, and temperature, and their relation to energy, radiation, and physical properties of matter. The behavior of these quantities is governed by the four laws of thermodynamics which convey a quantitative description using

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measurable
macroscopic physical
quantities, but may be
explained in terms of
microscopic
constituents by ...

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Thermal Conduction,
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of Gases, The First
and Second Laws of
Thermodynamics,
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zeroth and first law of
thermodynamics,
identify the
relationship and

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correct usage of
infinitesimal work,
work -energy,
heatcapacity, specific
heat, latent heat, and
enthalpy of a system

~~THERMODYNAMICS AND STATISTICAL PHYSICS~~ | Physics

Relation to heat and
internal energy. In
thermodynamics, heat
is energy in transfer to

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or from a statistical thermodynamic system, by mechanisms other than thermodynamic work or transfer of matter. Heat refers to a quantity transferred between systems, not to a property of any one system, or 'contained' within it. On the other hand, internal energy is a

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~~Thermal energy—
Wikipedia~~

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statistical mechanics
MCQs 1. The term
"thermodynamics"
comes from the Greek
words "therme" and
"dynamic" which
means _____. A.
Heat power B. Heat

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statistical theory, this
book presupposes no
exposure to statistics
or quantum
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