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The key to understanding electronic arrangement is
summarized in the Pauli exclusion principle: no two
electrons in an atom can have the same set of four
quantum numbers. This dramatically limits the number
of electrons that can exist in a shell or a subshell.

Organization of Electrons in Atoms – Introductory ...

You might have been taught that atoms are the smallest
particles on Earth, but inside of atoms are even smaller
things called electrons. ... The number of electrons in
an atom The way atoms bond ...

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